

Certificate of Analysis

BDG Synthesis certifies that this reference material meets or exceeds the specifications stated herein.

leil Beare

Neil Beare, PhD, Director 14 November 2014

Name: Vilanterol-d₄ Triphenylacetic Acid Salt

CAS Number: 503068-34-6 (unlabelled free base)

Structure:

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$$

Molecular Weight: $C_{24}H_{29}D_4Cl_2NO_5 \cdot C_{20}H_{16}O_2 = 778.79$

Lot Number: BDG 15131.2

Appearance: White, crystalline solid

Corrected Purity: 99.7 % (HPLC) - 0.4 % (diethyl ether) = 99.3 %

Isotopic Purity: Under 0.5 % d₀

Re-test Date: 14 November 2019

Storage and Handling: Temperature: refrigerate for prolonged storage; may be handled and shipped at

ambient temperature.

Humidity: not believed to be hygroscopic; may be handled in normal laboratory

atmosphere.

Light: protect from strong sunlight.

Caution: only experienced laboratory personnel should handle the material.

Version 1 (dd712) 1/5

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Identity and Purity

Proton NMR Spectrum

Identity: the signals are consistent with the proposed structure and in accord with literature where available. Isotopic Labelling: signals at the sites of deuteration are greatly diminished, compared with the spectrum of unlabelled material, indicating clean deuteration.

Residual Solvents: a small amount of diethyl ether (0.4 % w/w) is observed.

Impurities: no significant impurities are evident in the spectrum.

Carbon-13 NMR Spectrum

Identity: the signals are consistent with the proposed structure and in accord with literature where available. Isotopic Labelling: signals at the sites of deuteration have collapsed to small multiplets compared with the spectrum of unlabelled material, indicating clean deuteration.

High-resolution Mass Spectrum (ESI+)

Found m/z 490.2068. $C_{24}H_{30}D_4Cl_2NO_5$ [M+H]⁺ requires m/z 490.2065. The deviation of 0.6 ppm is within normally accepted limits for the establishment of identity by HRMS. No signal for d_0 material was seen (detection limit about 0.5 %).

HPLC

A sharp, symmetrical peak is observed (99.7 %). The peak at 19.0 minutes is identified as triphenylacetic acid. Note: in the absence of reference materials for preparing calibration curves, it is assumed that all peaks have the same detector response. Where possible, the conditions of analysis follow a pharmacopeial or literature method, or have been adapted from same.

Elemental Analysis

Found: C 67.87, H 5.87, D 1.04, N 1.71 % C₂₄H₂₉D₄Cl₂NO₅·C₂₀H₁₆O₂ Requires: C 67.86, H 5.82, D 1.03, N 1.80 %

The elemental analyses fall within generally accepted limits for establishing the molecular formula given. The results may also be taken to imply the absence of significant quantities of water or inorganic salts (which have not been elsewhere tested for because of sample size limitations).

The available quantity of custom-synthesised material is always small, and this limits the extent and type of analytical data which can be obtained. This Certificate is presented in descriptive format for use by analytical chemists who are trained in the use of custom-synthesised materials. Custom materials often contain higher levels of residual solvents and/or water, and we urge you to use the corrected purity where needed rather than the raw HPLC purity. This compound is intended for use as an analytical reference material and it is not for human administration. Structures are shown with relative stereochemistry unless otherwise specified.

The re-test date is assigned from experience gained with the material in the laboratory and/or on storage. It is not possible to perform formal storage studies because of the small amount of material available.

Proton NMR Spectrum of Vilanterol Triphenylacetic Acid Salt (top) and Vilanterol-d₄ Triphenylacetic Acid Salt (bottom) in Methanol-d₄

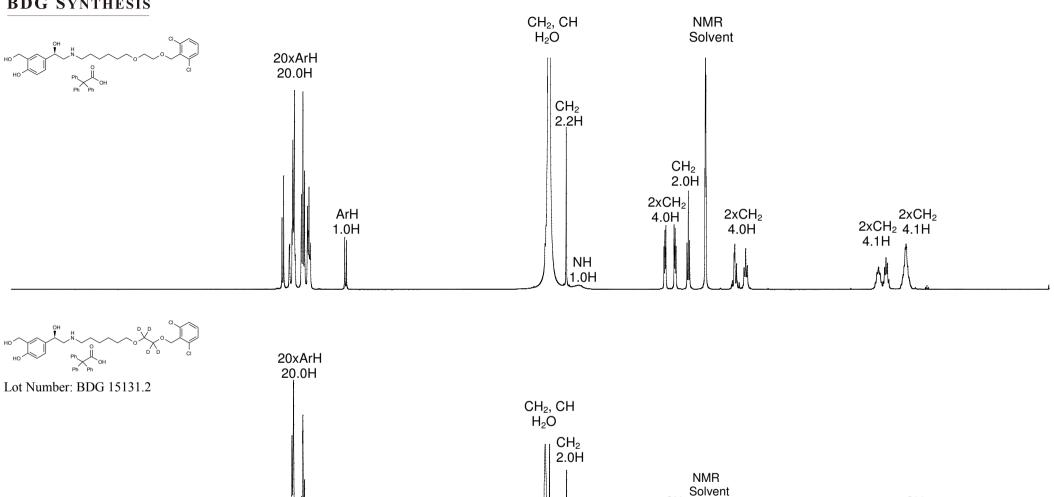
CH₂

2.1H

2xCH₂

4.0H

BDG SYNTHESIS



ArH

1.0H

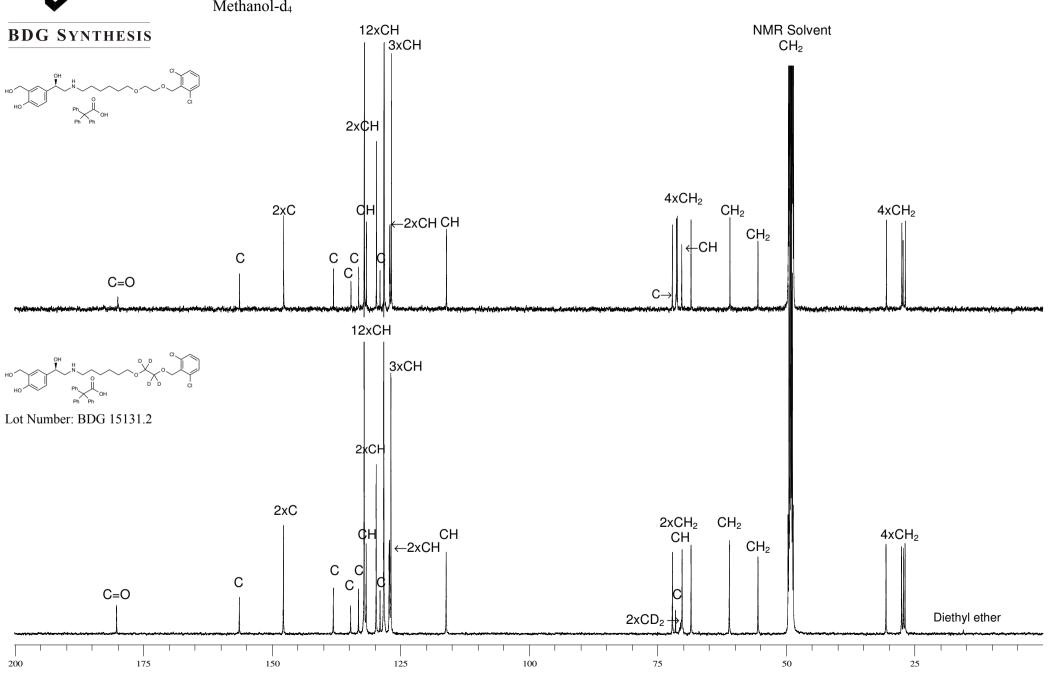
2xCH₂

Diethylether 0.4% w/w

2xCH₂

4.0H

Carbon-13 NMR Spectrum of Vilanterol Triphenylacetic Acid Salt (top) and Vilanterol-d₄ Triphenylacetic Acid Salt (bottom) in Methanol-d₄



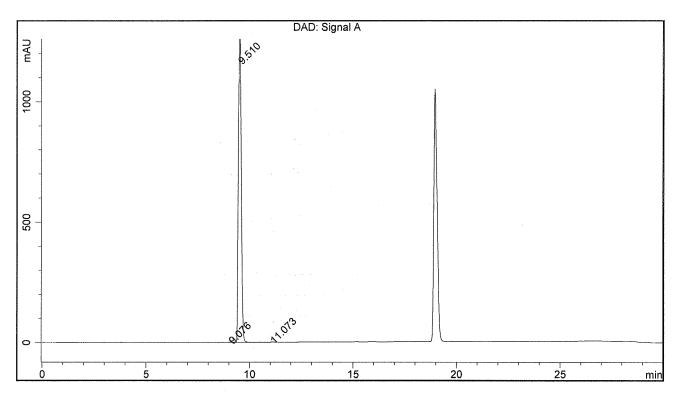
BDG - Analysis of Vilanterol-d4 Triphenylacetic Acid Salt

Column: Phenomenex Luna C18(2) 5um 250 x 4.6 mm
Guard: Phenomenex Security Guard C18 RP 4 x 3 mm
Mobile Phase A: 75:25 10 mM Potassium diHydrogen Phosphate pH=3.0: Acetonitrile
Mobile Phase B: 20:80 10 mM Potassium diHydrogen Phosphate pH=3.0: Acetonitrile
Gradient (A:B): T0=100:0, T20=0:100, T25=0:100, T27=100:0, T30=100:0

Column Temperature: 40 C

Flow Rate: 1.0 mL/min
Sample Solvent: 1:1 Mobile Phase A: Mobile Phase B:
Injection Volume: 10 uL
Detection: UV 220 nm

Sample Name	BDG 15131.2	Instrument	AnalyticalLC01
Acquisition	14/11/2014, 12:18:35	Method (rev.)	LC10633g (2)
Sequence	BDG_14Nov2014a	Vial Position	6
Operator	solvation010\cerityadmin	Injection	1 of 2



Area Percent Report

Peak#	RT	Peak Height	Peak Area	Width	Area %
1	9.08 min	1.6459	13.8214	0.1400 min	0.108 %
2	9.51 min	1288.3225	12743.1045	0.1594 min	99.739 %
3	11.07 min	1.6188	19.5287	0.1587 min	0.153 %